

Ignition of *n*-Hexane-Air Mixtures by Moving Hot Spheres

S. Coronel S. Menon R. Mével G. Blanquart J. E. Shepherd

Graduate Aeronautical Laboratories, California Institute of Technology

24th International Colloquium on the Dynamics of Explosions and Reactive Systems

National Central University
Taipei, Taiwan
July 28 - August 2, 2013



Summary

1. Motivation

- Accidental Ignition
- Hot Surface Ignition

2. Previous Work

3. Materials and Methods

4. Results

5. Conclusions/Future Work

Accidental Ignition

- Accidental ignition
 - electrostatic ignition of fuel
 - lightning strike
 - electrical faults in pumps, fuel quantity instrumentation
 - hot surface ignition
- Characterize fuel-oxidizer properties (*n*-hexane)
 - ignition delay time (Burcat et al. and Zhukov et al.)
 - heating rate on the low temperature oxidation of hexane by air (Boettcher et al.)
 - minimum ignition temperature (Boettcher)
 - minimum ignition energy (Bane)
 - laminar burning speed



TWA 800, NY 747-100, July 17, 1996



China Air Flight 120 caught fire in Okinawa Japan (BBC News, August 20, 2007)

Accidental Ignition

- Accidental ignition
 - electrostatic ignition of fuel
 - lightning strike
 - electrical faults in pumps, fuel quantity instrumentation
 - hot surface ignition
- Characterize fuel-oxidizer properties (*n*-hexane)
 - ignition delay time (Burcat et al. and Zhukov et al.)
 - heating rate on the low temperature oxidation of hexane by air (Boettcher et al.)
 - minimum ignition temperature (Boettcher)
 - minimum ignition energy (Bane)
 - laminar burning speed



TWA 800, NY 747-100, July 17, 1996



China Air Flight 120 caught fire in Okinawa Japan (BBC News, August 20, 2007)

Accidental Ignition

- Accidental ignition
 - electrostatic ignition of fuel
 - lightning strike
 - electrical faults in pumps, fuel quantity instrumentation
 - hot surface ignition
- Characterize fuel-oxidizer properties (*n*-hexane)
 - ignition delay time (Burcat et al. and Zhukov et al.)
 - heating rate on the low temperature oxidation of hexane by air (Boettcher et al.)
 - minimum ignition temperature (Boettcher)
 - minimum ignition energy (Bane)
 - laminar burning speed



TWA 800, NY 747-100, July 17, 1996



China Air Flight 120 caught fire in Okinawa Japan (BBC News, August 20, 2007)

Accidental Ignition

- Accidental ignition
 - electrostatic ignition of fuel
 - lightning strike
 - electrical faults in pumps, fuel quantity instrumentation
 - hot surface ignition

- Characterize fuel-oxidizer properties (*n*-hexane)

- ignition delay time (Burcat et al. and Zhukov et al.)
- heating rate on the low temperature oxidation of hexane by air (Boettcher et al.)
- minimum ignition temperature (Boettcher)
- minimum ignition energy (Bane)
- laminar burning speed



TWA 800, NY 747-100, July 17, 1996



China Air Flight 120 caught fire in Okinawa Japan (BBC News, August 20, 2007)

Accidental Ignition

- Accidental ignition
 - electrostatic ignition of fuel
 - lightning strike
 - electrical faults in pumps, fuel quantity instrumentation
 - hot surface ignition
- Characterize fuel-oxidizer properties (n -hexane)
 - ignition delay time (Burcat et al. and Zhukov et al.)
 - heating rate on the low temperature oxidation of hexane by air (Boettcher et al.)
 - minimum ignition temperature (Boettcher)
 - minimum ignition energy (Bane)
 - laminar burning speed



TWA 800, NY 747-100, July 17, 1996



China Air Flight 120 caught fire in Okinawa Japan (BBC News, August 20, 2007)

Accidental Ignition

- Accidental ignition
 - electrostatic ignition of fuel
 - lightning strike
 - electrical faults in pumps, fuel quantity instrumentation
 - **hot surface ignition**
- Characterize fuel-oxidizer properties (*n*-hexane)
 - ignition delay time (Burcat et al. and Zhukov et al.)
 - heating rate on the low temperature oxidation of hexane by air (Boettcher et al.)
 - minimum ignition temperature (Boettcher)
 - minimum ignition energy (Bane)
 - laminar burning speed

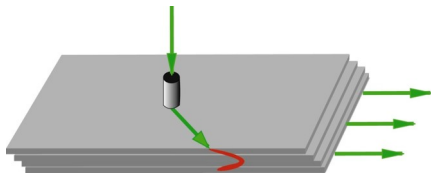


TWA 800, NY 747-100, July 17, 1996



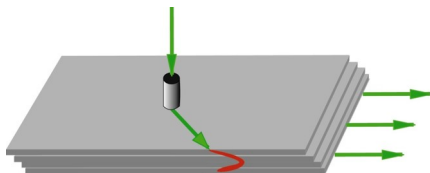
China Air Flight 120 caught fire in Okinawa Japan (BBC News, August 20, 2007)

Hot Surface Ignition



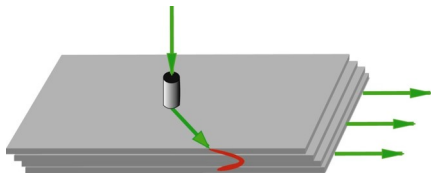
- Lightning attaching to conductor
- Current flows through composite material
- Heating of material
- Ejection of hot particles ($V_{particle} > 0$)

Hot Surface Ignition



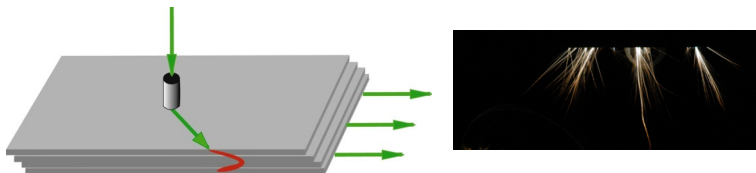
- Lightning attaching to conductor
- Current flows through composite material
- Heating of material
- Ejection of hot particles ($V_{particle} > 0$)

Hot Surface Ignition



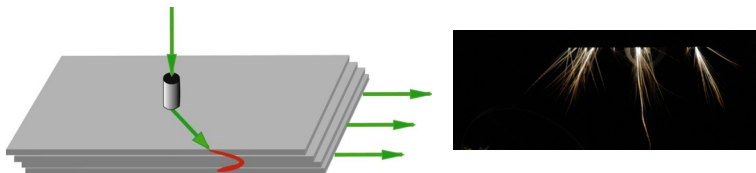
- Lightning attaching to conductor
- Current flows through composite material
- Heating of material
- Ejection of hot particles ($V_{particle} > 0$)

Hot Surface Ignition



- Lightning attaching to conductor
- Current flows through composite material
- Heating of material
- Ejection of hot particles ($V_{particle} > 0$)

Hot Surface Ignition



- Lightning attaching to conductor
- Current flows through composite material
- Heating of material
- Ejection of hot particles ($V_{particle} > 0$)

Ignition of *n*-hexane-air mixtures using moving hot particles

Summary

1. Motivation

2. Previous Work

- Experimental Setups
- Results

3. Materials and Methods

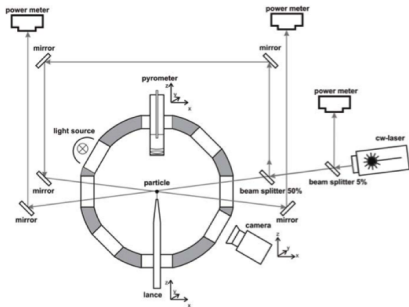
4. Results

5. Conclusions/Future Work

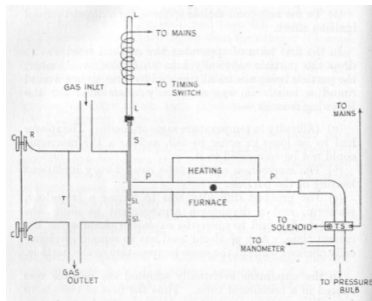
Experimental Setups : Stationary vs Moving Particles

Stationary particles heated via IR laser : Beyer et al. (2010), Dubaniewicz et al. (2000), Homan et al. (1981)

Moving heated particles introduced into a flammable environment : Silver (1937)

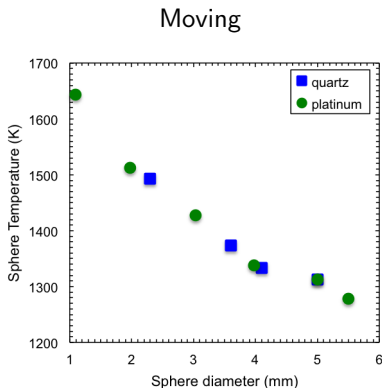
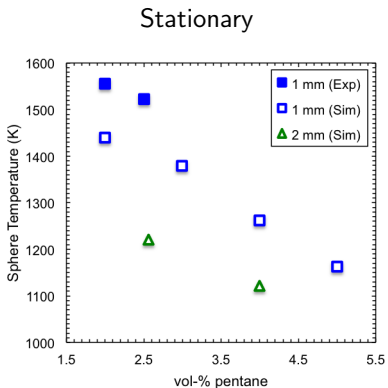


Experimental Setup, Beyer et al. (2010)



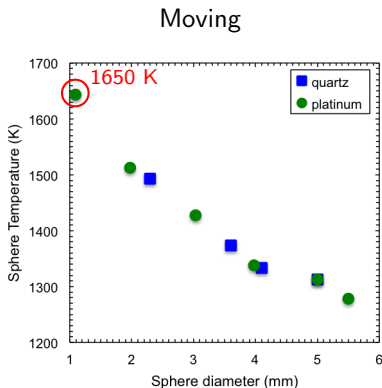
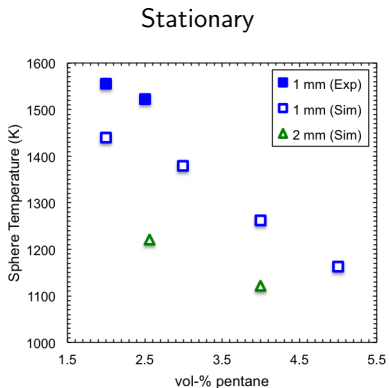
Experimental Setup, Silver (1937)

Results : Stationary vs Moving Particles



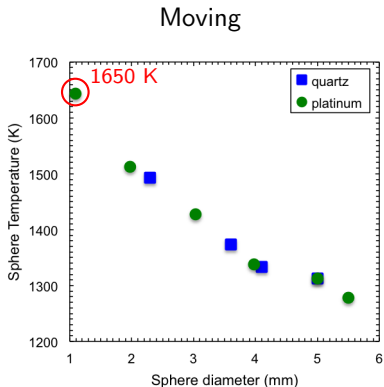
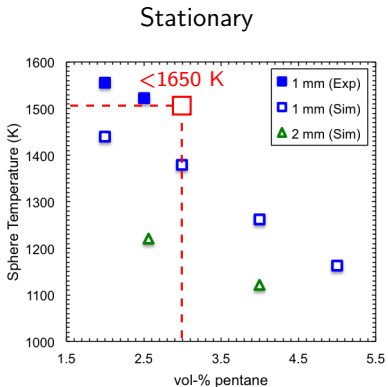
Pentane/air ignition results, Beyer et al. (2010) 3 vol-% pentane/air ignition, Silver (1937)

Results : Stationary vs Moving Particles



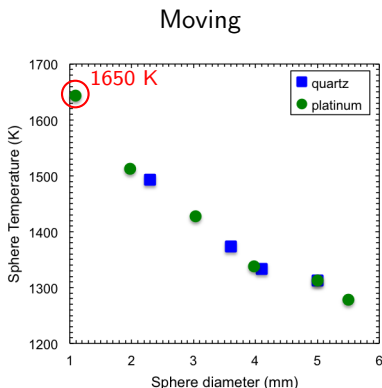
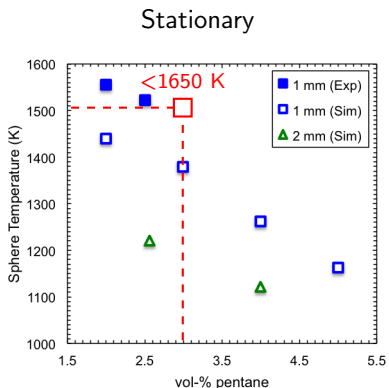
Pentane/air ignition results, Beyer et al. (2010) 3 vol-% pentane/air ignition, Silver (1937)

Results : Stationary vs Moving Particles



Pentane/air ignition results, Beyer et al. (2010) 3 vol-% pentane/air ignition, Silver (1937)

Results : Stationary vs Moving Particles

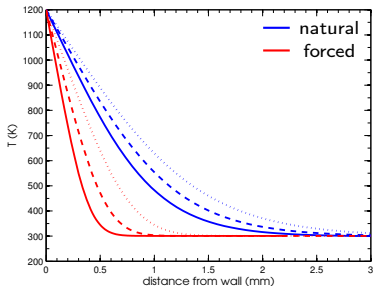
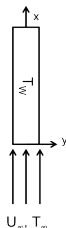


Pentane/air ignition results, Beyer et al. (2010) 3 vol-% pentane/air ignition, Silver (1937)

$$T_{stationary} < T_{moving}$$
$$D_{sphere} \uparrow T_{ignition} \downarrow$$

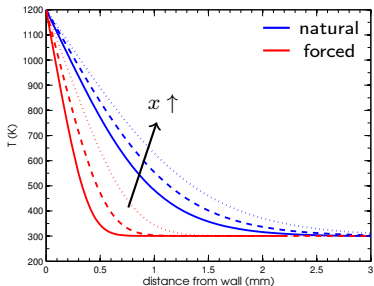
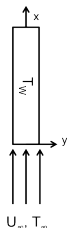
Natural and Forced Convection

- Parallel flow past a vertical plate :
 - $\rightarrow U_\infty = 0 \text{ m/s}$ and $U_\infty = 2.5 \text{ m/s}$
 - $\rightarrow T_\infty = 300 \text{ K}$
- Adiabatic plate : $T_w = 1200 \text{ K}$



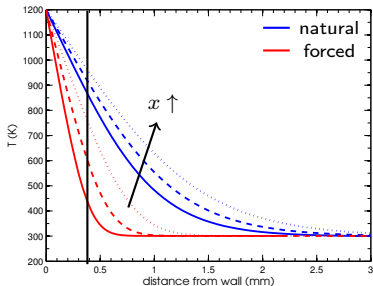
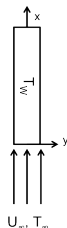
Natural and Forced Convection

- Parallel flow past a vertical plate :
 - $\rightarrow U_\infty = 0 \text{ m/s}$ and $U_\infty = 2.5 \text{ m/s}$
 - $\rightarrow T_\infty = 300 \text{ K}$
- Adiabatic plate : $T_w = 1200 \text{ K}$



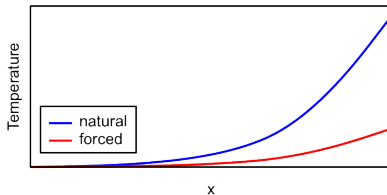
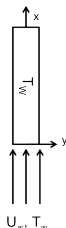
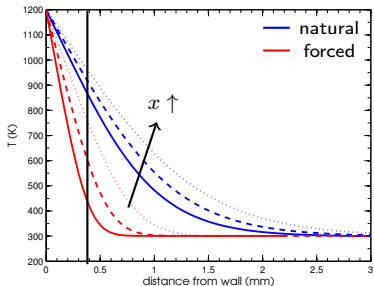
Natural and Forced Convection

- Parallel flow past a vertical plate :
 - $U_\infty = 0$ m/s and $U_\infty = 2.5$ m/s
 - $T_\infty = 300$ K
- Adiabatic plate : $T_w = 1200$ K



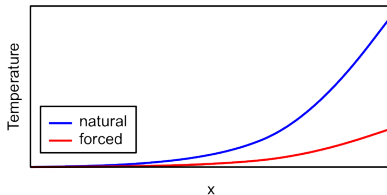
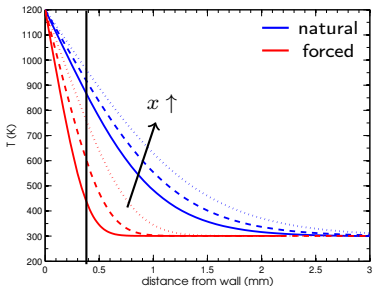
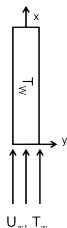
Natural and Forced Convection

- Parallel flow past a vertical plate :
 - $\rightarrow U_\infty = 0 \text{ m/s}$ and $U_\infty = 2.5 \text{ m/s}$
 - $\rightarrow T_\infty = 300 \text{ K}$
- Adiabatic plate : $T_w = 1200 \text{ K}$



Natural and Forced Convection

- Parallel flow past a vertical plate :
 - $\rightarrow U_\infty = 0 \text{ m/s}$ and $U_\infty = 2.5 \text{ m/s}$
 - $\rightarrow T_\infty = 300 \text{ K}$
- Adiabatic plate : $T_w = 1200 \text{ K}$

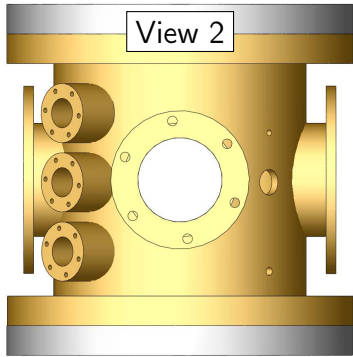
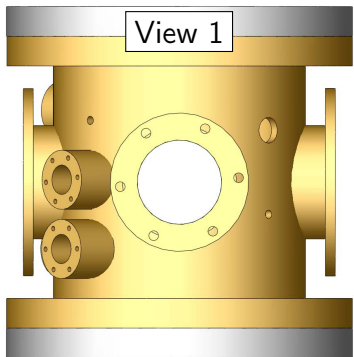


$$\rho c_p \left(u \frac{\partial T}{\partial x} + v \frac{\partial T}{\partial y} \right) - k \frac{\partial^2 T}{\partial y^2} = -q_c A e^{-\frac{E_a}{RT}}$$

Summary

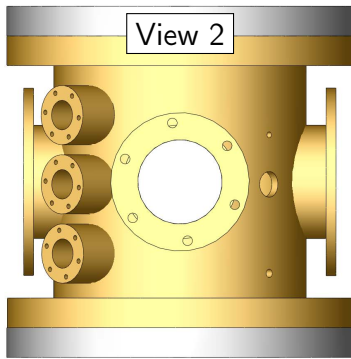
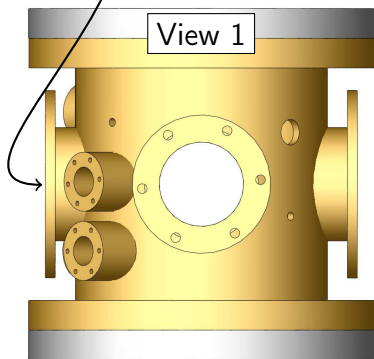
1. Motivation
2. Previous Work
- 3. Materials and Methods**
 - Experimental Setup
4. Results
5. Conclusions/Future Work

Experimental Setup : Combustion Vessel



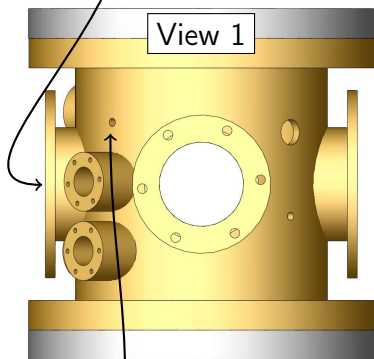
Experimental Setup : Combustion Vessel

11.7 cm diameter windows

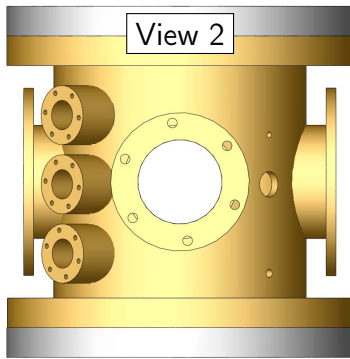


Experimental Setup : Combustion Vessel

11.7 cm diameter windows



pressure manometer



Experimental Setup : Combustion Vessel

11.7 cm diameter windows

vacuum

View 1

View 2

pressure manometer

Experimental Setup : Combustion Vessel

11.7 cm diameter windows

vacuum

View 1

View 2

pressure manometer

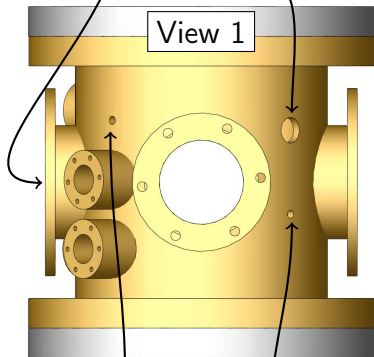
gas fill line

Experimental Setup : Combustion Vessel

11.7 cm diameter windows

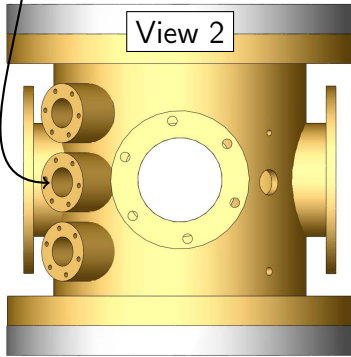
vacuum

View 1



fan mixer

View 2

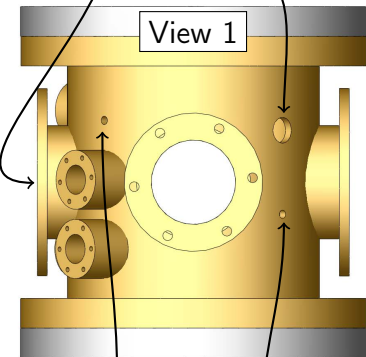


Experimental Setup : Combustion Vessel

11.7 cm diameter windows

vacuum

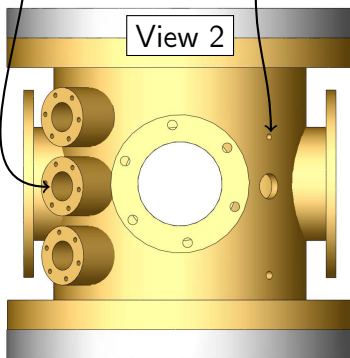
View 1



fan mixer

septum

View 2

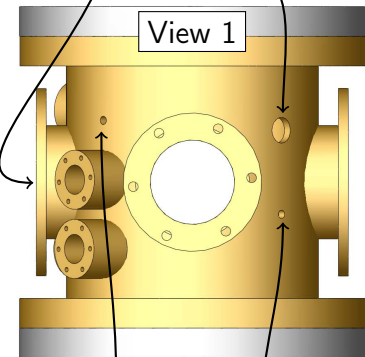


Experimental Setup : Combustion Vessel

11.7 cm diameter windows

vacuum

View 1



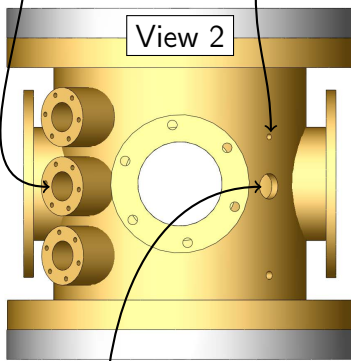
pressure manometer

gas fill line

fan mixer

septum

View 2



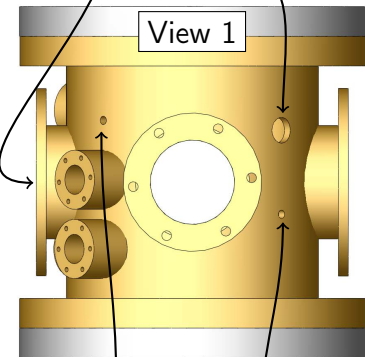
piezoresistive pressure transducer

Experimental Setup : Combustion Vessel

11.7 cm diameter windows

vacuum

View 1



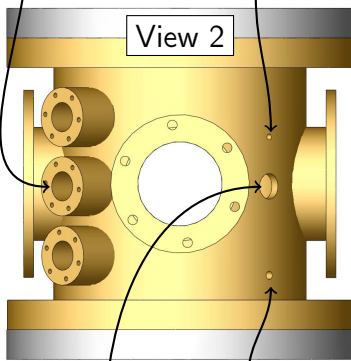
pressure manometer

gas fill line

fan mixer

septum

View 2

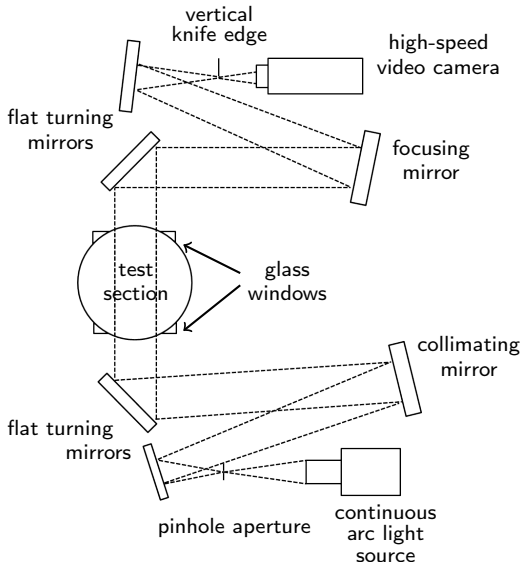


thermocouple

piezoresistive pressure transducer

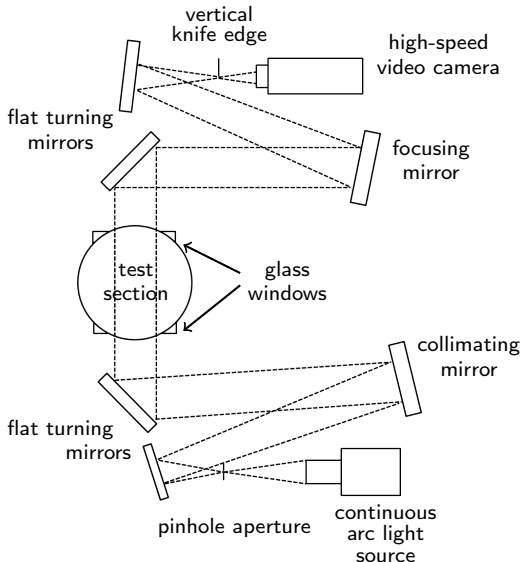
Experimental Setup : Schlieren Setup

- Observe changes in the density gradient of the fluid due to variations in the refractive index
- Visualize flame :
 - very hot flame propagating into cold unburned reactants
- High speed camera :
 - 10,000 frames per second
 - 512×512 resolution



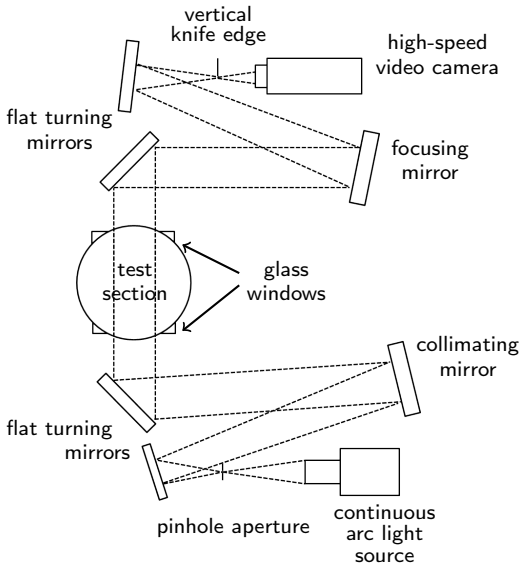
Experimental Setup : Schlieren Setup

- Observe changes in the density gradient of the fluid due to variations in the refractive index
- Visualize flame :
 - very hot flame propagating into cold unburned reactants
- High speed camera :
 - 10,000 frames per second
 - 512×512 resolution



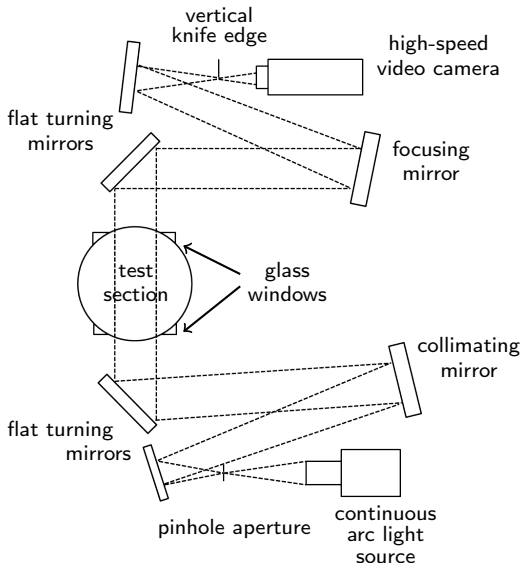
Experimental Setup : Schlieren Setup

- Observe changes in the density gradient of the fluid due to variations in the refractive index
- Visualize flame :
 - very hot flame propagating into cold unburned reactants
- High speed camera :
 - 10,000 frames per second
 - 512×512 resolution



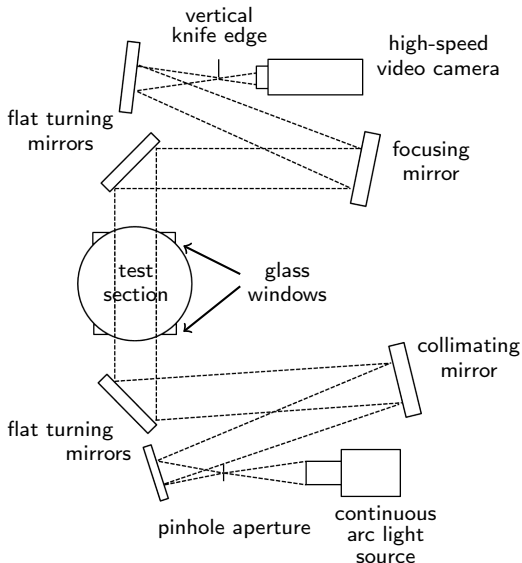
Experimental Setup : Schlieren Setup

- Observe changes in the density gradient of the fluid due to variations in the refractive index
- Visualize flame :
 - very hot flame propagating into cold unburned reactants
- High speed camera :
 - 10,000 frames per second
 - 512×512 resolution



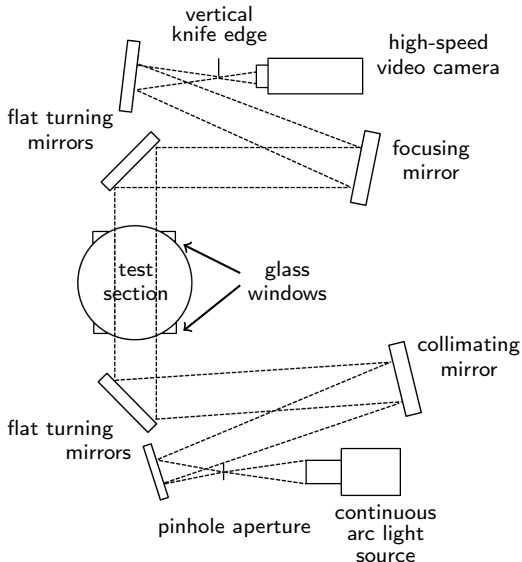
Experimental Setup : Schlieren Setup

- Observe changes in the density gradient of the fluid due to variations in the refractive index
- Visualize flame :
 - very hot flame propagating into cold unburned reactants
- High speed camera :
 - 10,000 frames per second
 - 512×512 resolution

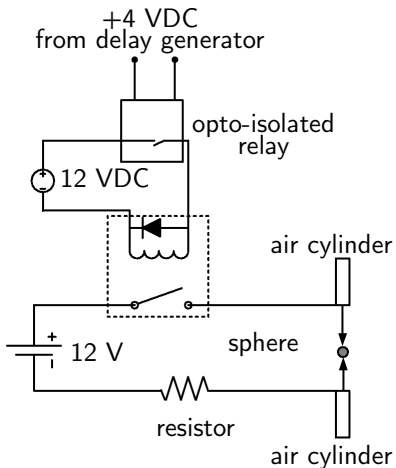


Experimental Setup : Schlieren Setup

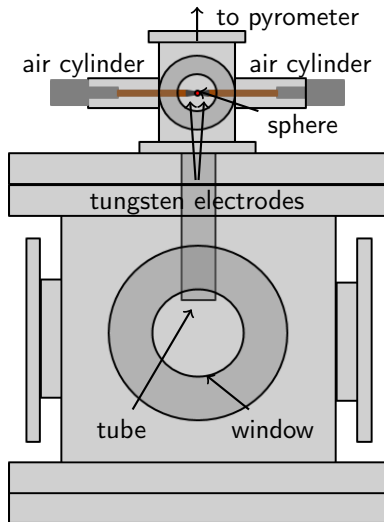
- Observe changes in the density gradient of the fluid due to variations in the refractive index
- Visualize flame :
 - very hot flame propagating into cold unburned reactants
- High speed camera :
 - 10,000 frames per second
 - 512×512 resolution



Experimental Setup : High Current Particle Heater

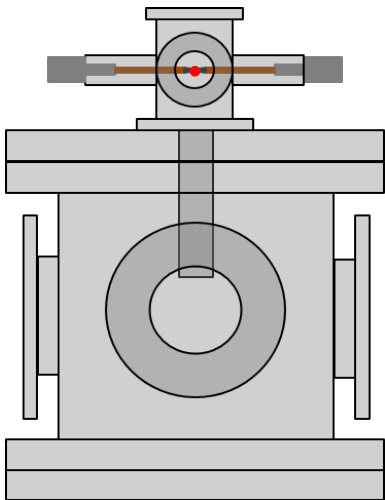


Heating circuit schematic

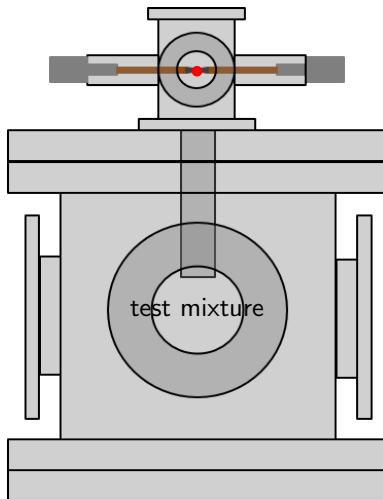


Particle heater chamber (top) and combustion vessel (bottom)

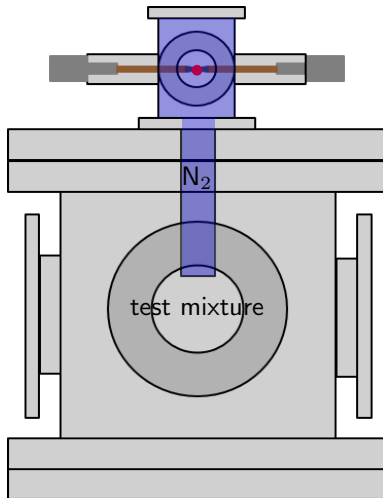
Experimental Setup : High Current Particle Heater



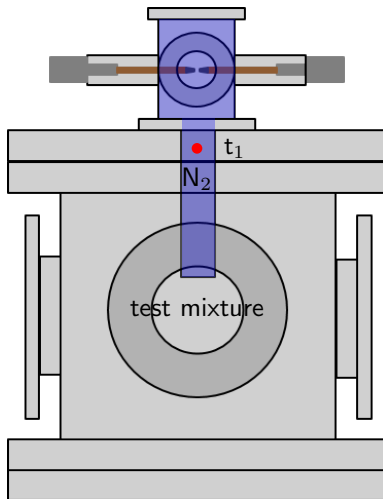
Experimental Setup : High Current Particle Heater



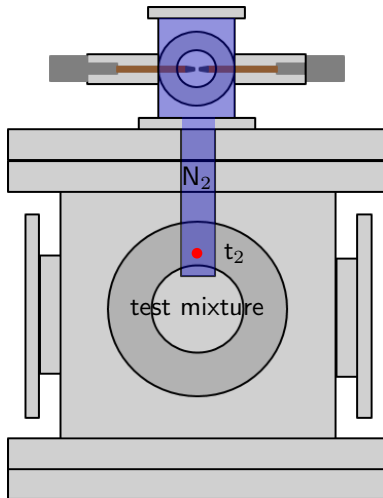
Experimental Setup : High Current Particle Heater



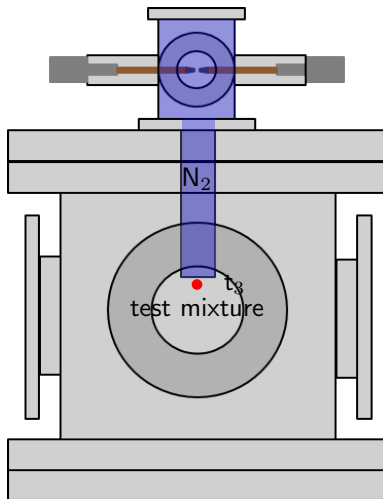
Experimental Setup : High Current Particle Heater



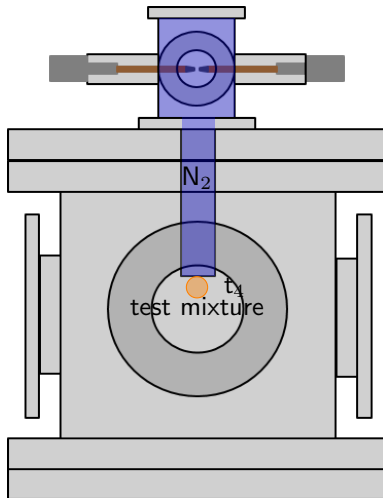
Experimental Setup : High Current Particle Heater



Experimental Setup : High Current Particle Heater



Experimental Setup : High Current Particle Heater

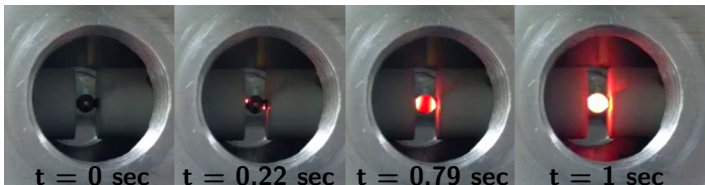


Summary

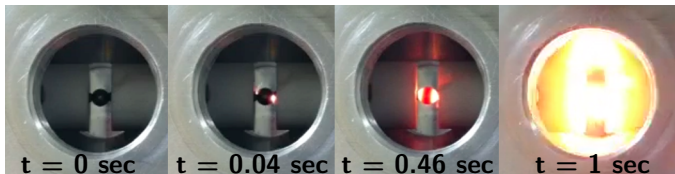
1. Motivation
2. Previous Work
3. Materials and Methods
- 4. Results**
 - Heating in Air
 - Ignition Results
 - Simulation Results
 - Comparison
5. Conclusions/Future Work

Results : High Current Heating in Air

Heating in air, $R = 17 \text{ m}\Omega$, $d = 4 \text{ mm}$, heating time $\approx 1 \text{ sec}$



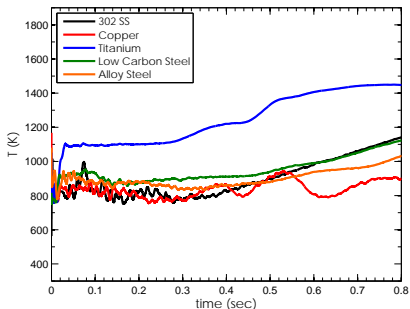
Titanium (Ti-6Al-4V) $\rightarrow T_{max} \approx 1200 \text{ K}$



Titanium (Ti-6Al-4V) $\rightarrow T_{max} \approx 1300 \text{ K}$

Results : High Current Heating in Air

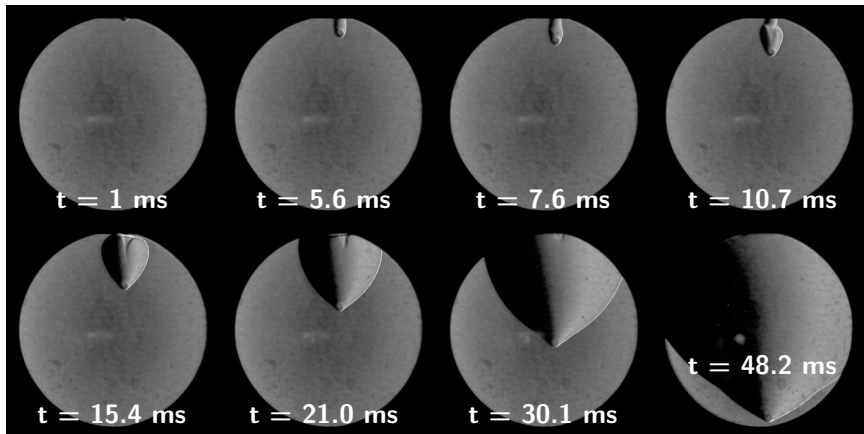
Heating in air, $R = 13 \text{ m}\Omega$, $d = 4 \text{ mm}$



Material	$T_{melting}$ (K)
Copper	1350
Carbon steel, alloy steel, 302 SS	1700-1800
Titanium	1950

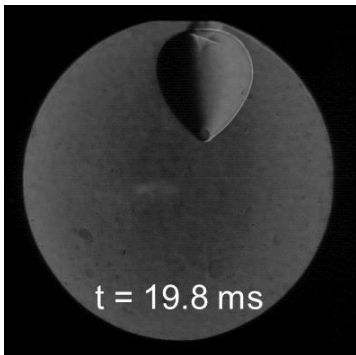
Results : Ignition Test

$$T_{surface} = 1199 + 42 \text{ K}/-10 \text{ K}$$



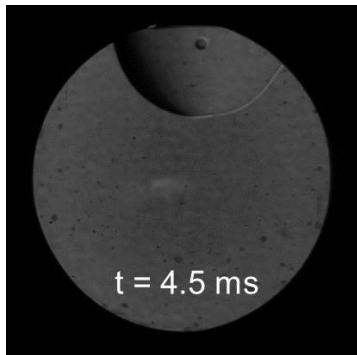
Results : Flame Geometry

n-Hexane–Air, $\Phi = 0.9$
 $P = 100$ kPa



$$V_S^0 = 2.6 \text{ m/s}$$
$$V_P \approx 2.3 \text{ m/s}$$

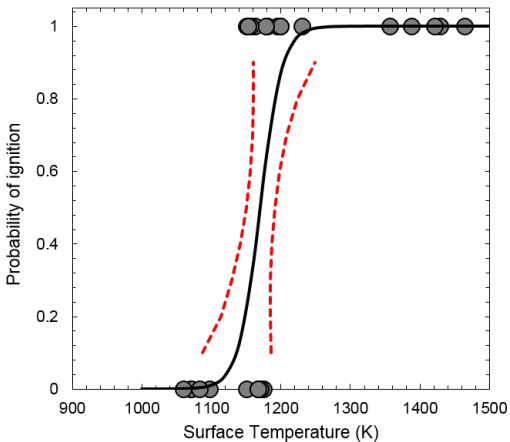
n-Hexane–O₂–40% N₂, $\Phi = 0.9$
 $P = 50$ kPa



$$V_S^0 = 23 \text{ m/s}$$
$$V_P \approx 2.3 \text{ m/s}$$

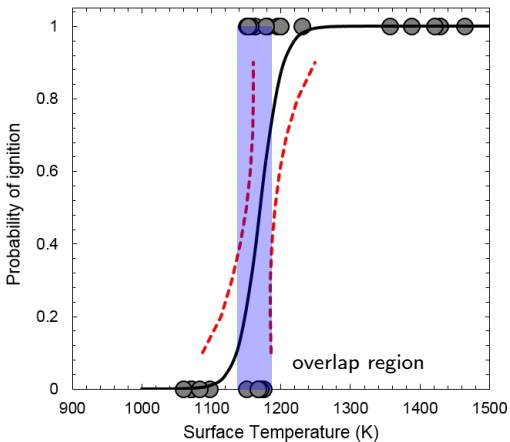
Results : Ignition Tests

Probability distribution : $P_0 = 100$ kPa, $T_0 = 300$ K and $\Phi = 0.9$
of tests : 26



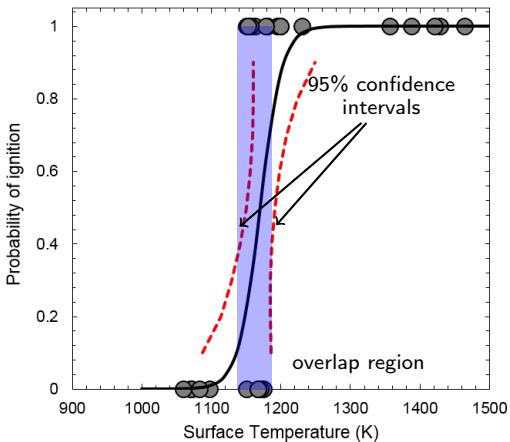
Results : Ignition Tests

Probability distribution : $P_0 = 100$ kPa, $T_0 = 300$ K and $\Phi = 0.9$
of tests : 26



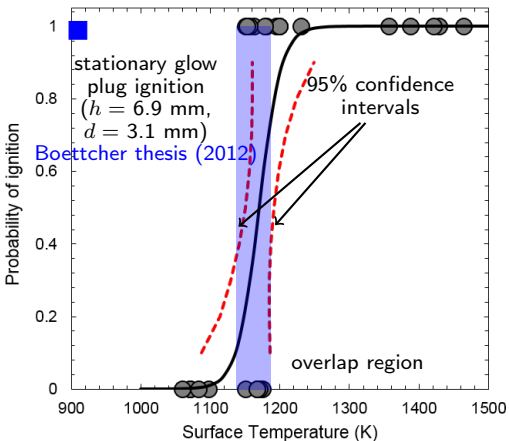
Results : Ignition Tests

Probability distribution : $P_0 = 100$ kPa, $T_0 = 300$ K and $\Phi = 0.9$
of tests : 26



Results : Ignition Tests

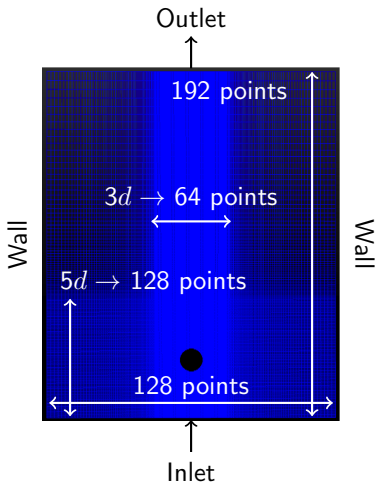
Probability distribution : $P_0 = 100$ kPa, $T_0 = 300$ K and $\Phi = 0.9$
of tests : 26



Simulation Setup

- Grid
 - Axisymmetric
 - Locally refined around stationary hot particle
 - 192x128 mesh points
- Boundary conditions
 - Constant particle temperature
 - Wall temperature : 300 K
 - Inert surfaces
 - Neumann boundary condition for species
- Initial conditions
 - $P_0 = 100$ kPa, $T_0 = 300$ K and $\Phi = 0.9$
 - Flow N_2 at $t = 0-200$ ms and n -heptane¹-air at $t > 200$ ms

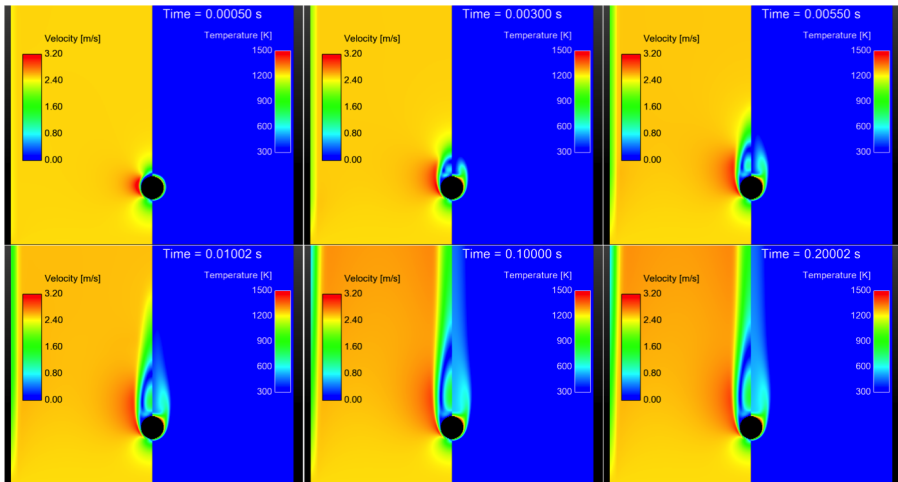
¹ Blanquart G., Pepiot-Desjardins P., and Pitsch H. (2009). *Combustion and Flame*. 156 : 588-607.



- Low Mach number Navier-Stokes equations and detailed chemistry calculations

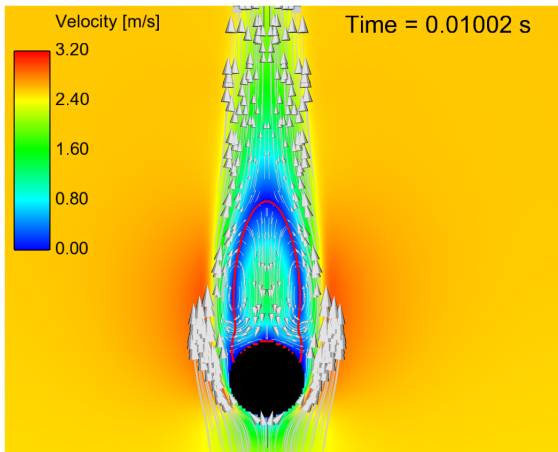
Results : Simulations (N_2 Hot Wake)

N_2 velocity and thermal boundary layers, stabilized at $t > 100$ ms,
 $T_{surface} = 1500$ K

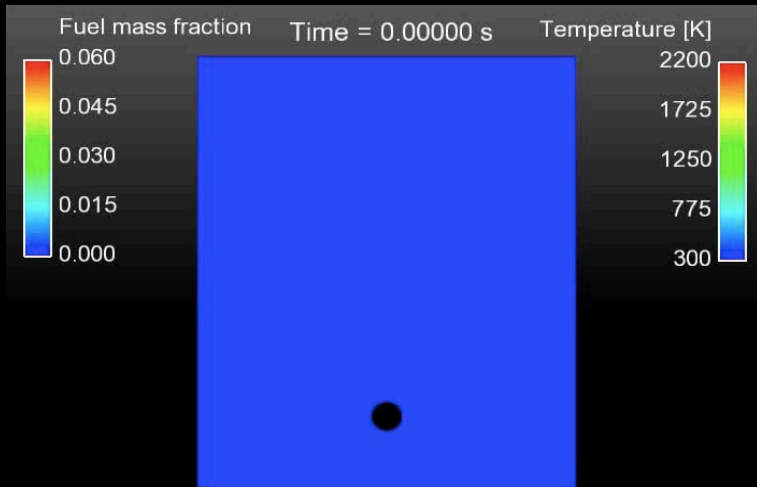


Results : Simulations (N₂ Hot Wake)

$$T_{surface} = 1500 \text{ K}$$

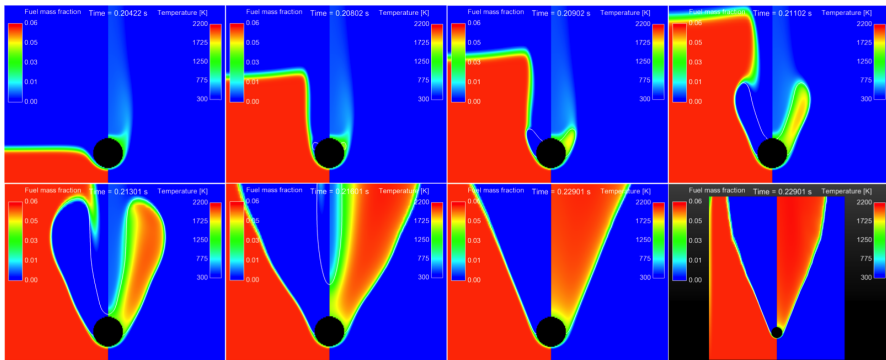


n -Heptane-air : $P_0 = 100$ kPa, $T_0 = 300$ K, $\Phi = 0.9$
 $d = 4$ mm, $V = 2.5$ m/s, $T_{surface} = 1450$ K



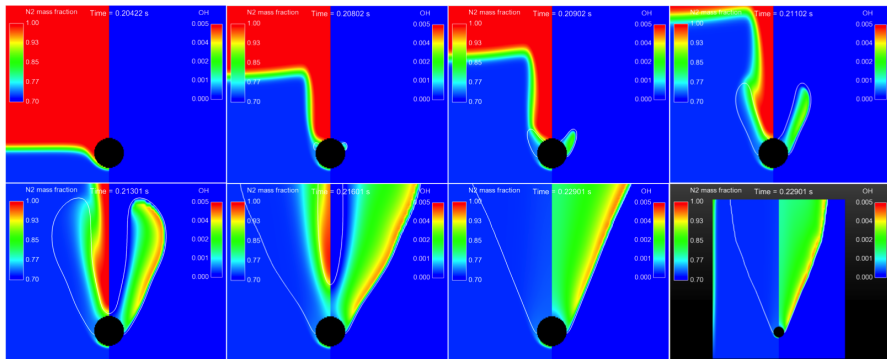
Results : Ignition Simulations

$T_{surface} = 1450$ K, fuel mass fraction and temperature



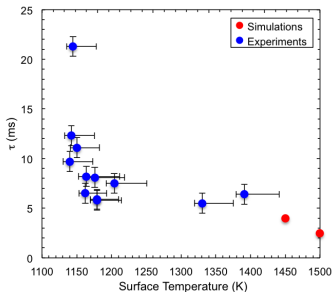
Results : Ignition Simulations

$$T_{surface} = 1450 \text{ K}, N_2 \text{ mass fraction and OH}^*$$



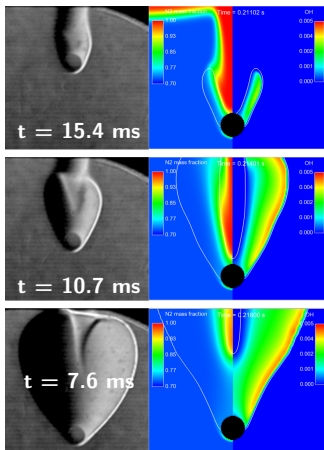
Results : Experiments and Simulations

Ignition delay time



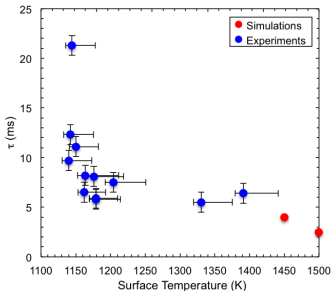
- Ignition delay time ≈ 5 ms
- Increase in ignition delay time while approaching minimum ignition temperature

Experiment : $T_{surface} = 1199$ K,
Simulation : $T_{surface} = 1450$ K



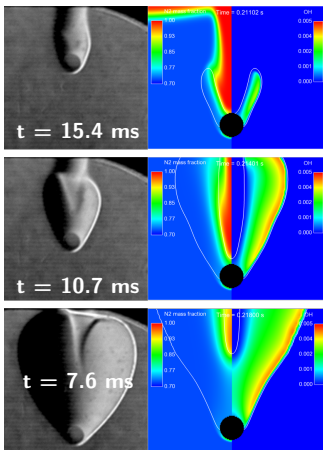
Results : Experiments and Simulations

Ignition delay time



- Ignition delay time ≈ 5 ms
- Increase in ignition delay time while approaching minimum ignition temperature

Experiment : $T_{surface} = 1199$ K,
Simulation : $T_{surface} = 1450$ K



Summary

1. Motivation
2. Previous Work
3. Materials and Methods
4. Results
5. Conclusions/Future Work

Conclusions/Future Work

Conclusions

- Ignition of *n*-hexane–air mixtures using moving heated particles ($P_0 = 100$ kPa, $T_0 = 300$ K, $\Phi = 0.9$) with $V_P \approx 2.3$ m/s
 - 50% probability of ignition with $T_{surface} = 1170$ K using 4 mm diameter titanium spheres
 - Interaction of flame and sphere for mixtures with flame speeds comparable to the particle velocity
- Qualitative agreement between experiments and simulation
 - Nitrogen recirculation region that leads to a particular flame propagation behaviour
 - Ignition close to sphere surface near flow separation region
- Comparable ignition delay times between experiments and simulations
 - 35-60% difference at $T_{surface} > 1400$ K

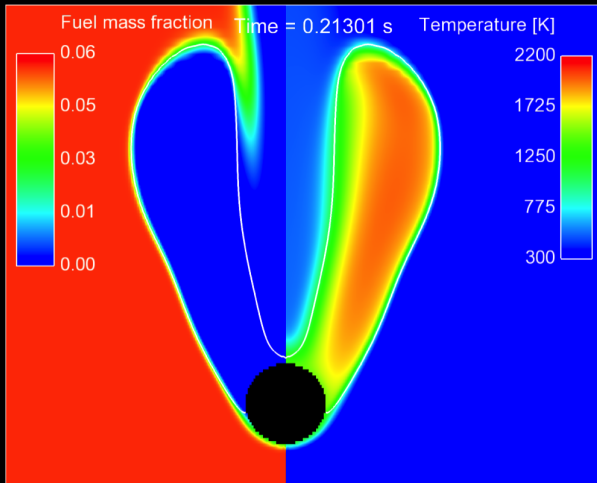
Future Work

- New particle heating method : CO_2 laser
 - Repeatable surface temperatures
 - Use of metallic or non-metallic materials
 - Easy switch to larger or smaller particles

Acknowledgements

The present work was carried out in the Explosion Dynamics Laboratory of the California Institute of Technology and was supported by The Boeing Company through a Strategic Research and Development Relationship Agreement CT-BA-GTA-1.

Thank You



David Marxer photography