

# Reduction Methodology for Detailed Kinetic Mechanisms: Application to *n*-Hexane-Air Hot Surface Ignition

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Caltech



清华大学  
Tsinghua University

# Thermal Ignition Hazards

**Motivation:** understand thermal ignition hazards present in the aviation, nuclear, mining, and manufacturing sectors.



Frictional sparks and hot spots<sup>1</sup>

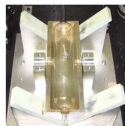


China Air flight 120, 2007



TWA flight 800, 1996

**Previous work:** extensive work has been performed at Caltech in the context of aviation safety using *n*-hexane as a surrogate for kerosene.

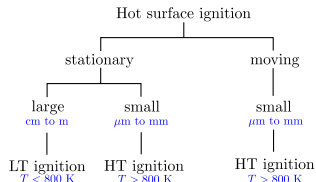


<sup>1</sup>S. Haworth et al. Symposium Series No. 150, 2004

# Numerical Predictions

**Problem:** experimental testing is time consuming

- Large number of hot surface geometries and mixture conditions



**Solution:** perform multi-dimensional transient numerical simulations.

- Predictive simulations require the use of detailed chemical mechanisms to predict the ignition and flame propagation events
  - *n*-Hexane: 531 species and 2628 reactions
  - Not feasible for modeling real engineering tests or applications

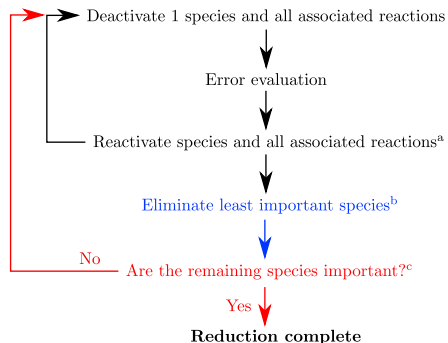
**How can computational time be reduced? Mechanism reduction!**

# Kinetic Mechanism Reduction

Use modified version of reduction mechanism employed by Davidenko et al.<sup>2</sup> using one target phenomenon: **auto-ignition process**

**Evaluated parameters ( $q$ ):** time to peak thermicity, maximum thermicity, equilibrium temperature, equilibrium mean molar mass, and temporal profiles of, thermicity, temperature, and mean molar mass.

- $s_q = \epsilon_q / \epsilon_q^*$  where  $\epsilon_q = |q - q^{\text{ref}}| / q^{\text{ref}}$   $q$ : reduced mechanism,  $q^{\text{ref}}$ : complete mechanism,  $\epsilon_q^*$ : error limit; calculate  $S = \sum s_q$  across all conditions.
- ~~$S^1 < S^2 < \dots < S^k$~~  where  $k$  is species deactivated
- $S = S^*$  where  $S^*$ : error limit



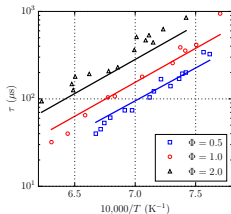
**Final mechanism: 62 species and 223 reactions**

<sup>2</sup>D. Davidenko, R. Mével, and G. Dupré, in Proceedings of the European Combustion Meeting, 2009.



# Mechanism Validation (1/3)

## Ignition delay time (constant volume adiabatic reactor)



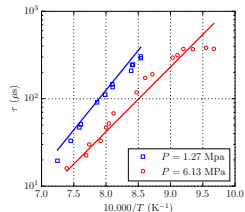
Mével et al.<sup>3</sup>:  $P = 350$  kPa;  
 $Y_{Ar} = 0.96$  *n*-hexane- $O_2$ -Ar

**RMSE**

□ 29  $\mu s$

○ 131  $\mu s$

△ 132  $\mu s$



Data from Zhukov et al.<sup>4</sup> for  
*n*-hexane-air at  $\Phi = 0.5$

**RMSE**

□ 47  $\mu s$

○ 107  $\mu s$

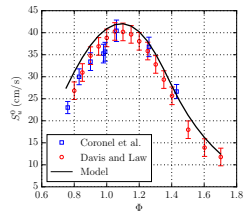
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<sup>3</sup>R. Mével, U. Niedzielska, J. Melguizo-Gavilanes, S. Coronel, and J. E. Shepherd, *Combustion Science and Technology*, 188:2267-2283, 2016.

<sup>4</sup>V. P. Zhukov, V. A. Sechenov, and A. Y. Starikovskii, *Combustion and Flame*, 136:257-259, 2004.

# Mechanism Validation (2/3)

## Laminar burning speed (1-D freely propagating flame)



Coronel et al.<sup>5</sup> and, Davis and Law<sup>6</sup>:  $P = 100$  kPa,  $T = 300$  K  
*n*-hexane-air

RMSE

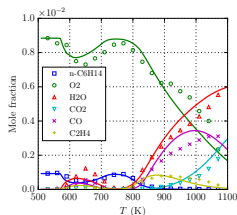
□ 3.6 cm/s

○ 2.5 cm/s

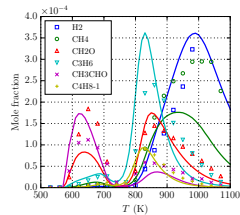
<sup>5</sup>S. Coronel, S. Lapointe, R. Mével and J. E. Shepherd, in preparation for Fuel.

<sup>6</sup>S. G. Davis and C. Law C, Combustion Science and Technology, 140:427-449, 1998.

## Mechanism Validation (3/3)

Jet-stirred reactor<sup>7</sup> (perfectly-stirred reactor-PSR)

Major species



Minor species

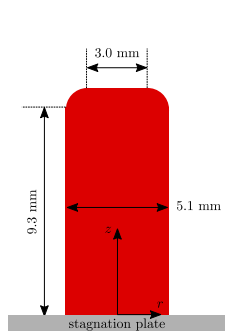
## Normalized root-mean-square error

Species	NRMSE	Species	NRMSE
<i>n</i> -C <sub>6</sub> H <sub>14</sub>	0.35	H <sub>2</sub>	0.5
O <sub>2</sub>	0.15	CH <sub>4</sub>	1.4
H <sub>2</sub> O	0.48	CH <sub>2</sub> O	0.4
CO <sub>2</sub>	0.7	C <sub>3</sub> H <sub>6</sub>	1.2
CO	1.2	CH <sub>3</sub> CHO	0.6
C <sub>2</sub> H <sub>4</sub>	10.6	C <sub>4</sub> H <sub>8</sub> -1	0.4

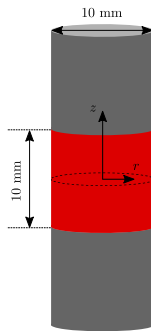
<sup>7</sup>K. Zhang, C. Banyon, C. Togbé, P. Dagaut, J. Bugler, and H. J. Curran, *Combustion and Flame*, 162:4194-4207, 2015.

# Multi-Dimensional Numerical Calculations

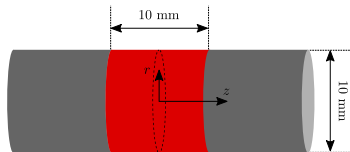
## Hot surface geometries



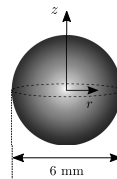
Commercial glow plug  
stationary



Vertical cylinder  
stationary



Horizontal cylinder  
stationary



Sphere  
moving

# Mesh Geometry Setup

## Mesh geometry

	Glow plug	Vertical cylinder	Horizontal cylinder	Sphere
Config.	2D axisymmetric about $z$	2D axisymmetric about $z$	2D planar normal to $z$	2D axisymmetric about $z$
Cells	180,000	80,000	60,000	300,000
Cell size ( $\mu\text{m}$ )	80	40	40	60

# Numerical Solution

- Variable-density, reactive Navier-Stokes
- Temperature dependent transport and thermodynamic properties
- Differential diffusion included using constant non-unity Lewis numbers
- *n*-Hexane-air:  $\Phi = 0.9$ ,  $T_0 = 300$  K, and  $P_0 = 100$  kPa

Temperature ramp: 220 K/s

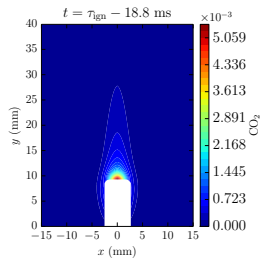
Temperature ramp: 220 K/s

Temperature ramp: 220 K/s

Temperature ramp and then drop

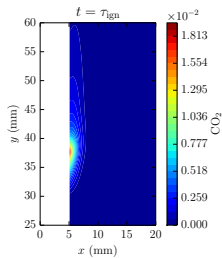
# CO<sub>2</sub> Mass Fraction

## Glow plug



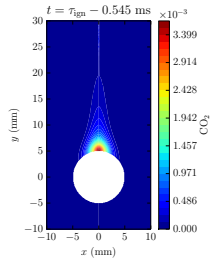
Ignition occurs above glow plug in stagnation region

## Vertical cylinder



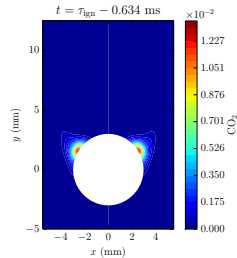
Ignition occurs along the top limit of the hot surface

## Horizontal cylinder



Ignition occurs in the rear stagnation point

## Moving sphere



Ignition occurs in region of flow separation

# Ignition Predictions

## Minimum surface temperature required for ignition

Geometry	Exp. (K)	Calc. (K)	$\Delta T/T_{\text{exp}}$	% Diff.
Hor. Cyl <sup>8</sup>	1180	1093	0.07	7%
Vert. Cyl. <sup>8</sup>	1270	1191	0.06	6%
Sphere <sup>9</sup>	1224	1300	0.06	6%
Glow Plug <sup>10</sup>	1275	1162	0.09	9%

Experimental and numerical results for all geometries tested

<sup>8</sup>L. Boeck, M. Meijers, A. Kink, R. Mével, and J. E. Shepherd, *Combustion and Flame*, 185:265-277, 2017.

<sup>9</sup>S. A. Coronel, J. Melguizo-Gavilanes, R. Mével, and J. E. Shepherd, submitted to *Combustion and Flame*.

<sup>10</sup>J. Melguizo-Gavilanes, A. Nové-Josserand, S. A. Coronel, R. Mével, and J. E. Shepherd, *Combustion Science and Technology*, 188:2060-2076, 2016



# Conclusions

- The reduced mechanism performs well when applied to 1D freely propagating flames
- Ignition predictions within 10% of measured experimental thresholds
  - Natural and forced convection
  - Flows with stagnation and separation points
  - Do not account for surface chemistry
- Largest temperature difference observed for commercial glow plug (9%)
  - Surface temperature inhomogeneities observed experimentally
  - Experimental measurement is made at the top edge of glow plug and not at ignition location
- Significant decrease in computational time
  - Maybe Josue can put a rough estimate of the savings in computational time?
- Rigorous validation of reduced mechanism and solver indicates that simulations of real-engineering geometries and flow configurations can be performed